## **Free Study Guide For Chemistry**

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questions: Atoms reaction, bonding nature and properties, chemical bonds, intermolecular forces, and types of bonds.

The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon, as it forms around particles of a metal catalyst. Carbon nanotubes are a recently discovered allotrope of carbon. Three other allotropes of carbon-buckyballs, graphite, and diamond-are illustrated at the left, as is the molecule methane, CH4, from which nanotubes and buckyballs can be made. The element carbon forms an amazing number of compounds with structures that follow from simple methane, found in natural gas, to the complex macromolecules that serve as the basis of life on our planet. The study of chemistry also follows from the simple to the more complex, and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement. From acids to alloys and equations to evaporation, this guide makes complex topics easy to grasp at a glance. Perfect support for coursework, homework, and exam revision. Each topic is fully illustrated, to support the information, make the facts crystal clear, bring the science to life and make studying a breeze. A large central image explains the idea visually and each topic is summed up on a single page, helping children to quickly get up to speed and really understand how chemistry works. For key ideas, "How it Works" and "Look Closer" boxes explain the theory with the help of simple graphics. And for revision, a handy "Key Facts" box provides a simple summary you can check back on later. With clear, concise coverage of all the core topics, Super Simple Chemistry is the perfect accessible guide to chemistry for children, supporting classwork, and making studying for exams the easiest it's ever been. Revised with the assistance of new author Christopher Hadad, Ohio State University the Study Guide and Solutions Manual offers solutions to both in-text and end-of-chapter problems with an explanation of the answers. For each chapter, the guide provides a chapter summary, learning objectives, and a summary of synthetic methods and reaction mechanisms. Review problems on synthesis and 100 multiple-choice sample test questions offer students additional problem-solving practice.

Study Guide to Accompany Calculus for the Management, Life, and Social Sciences

A Self-Study Guide to the Principles of Organic Chemistry: Key Concepts, Reaction Mechanisms, and Practice Questions for the Beginner will help students new to organic chemistry grasp the key concepts of the subject quickly and easily, as well as build a strong foundation for future study. Starting with the definition of "atom," the author explains molecules, electronic configuration, bonding, hydrocarbons, polar reaction mechanisms, stereochemistry, reaction varieties, organic spectroscopy, aromaticity and aromatic reactions, biomolecules, organic polymers, and a synthetic approach to organic compounds. The over one hundred diagrams and charts contained in this volume will help students visualize the structures and bonds as they read the text, and make the logic of organic chemistry clear and easily understood. Each chapter ends with a list of frequently-asked questions and answers, followed by additional practice problems. Answers are included in the Appendix.

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